

1,3-DIPOLAR CYCLOADDITION OF PYRAZOLID-3-ONE AZOMETHINIMINES AND E- β -NITRO-
STYRENE - ELUCIDATION OF THE GENERATION OF SEEMINGLY "NON CISOID" ADDUCTS

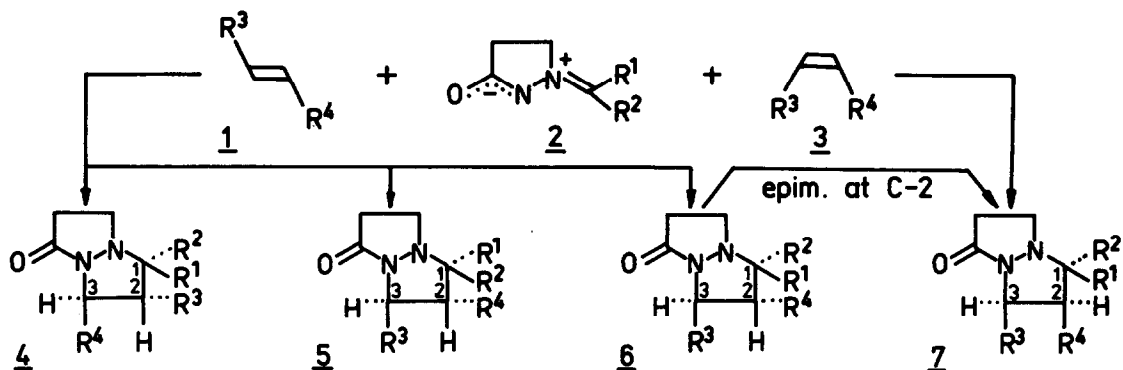
H e l m u t D o r n

Zentralinstitut für Organische Chemie , Akademie der Wissenschaften der DDR
DDR 1199 Berlin - Adlershof , Rudower Chaussee 5 , DDR

Abstract : Highly polar 1,3-dipoles like pyrazolid-3-one azomethinimines stereospecifically add polar dipolarophiles . A suggested "non cisoid" adduct of E- β -nitrostyrene is generated by epimerization at C-2 , catalysed by Kieselgel 40 MERCK a n d small amounts of pyrazolid-3-one .

Are highly polar 1,3-dipoles like pyrazolid-3-one azomethinimines 2¹ [$\mu(\underline{2a})=6.80 \pm 0.03$ D , dioxane ²] able to add polar dipolarophiles in polar solvents [E_T values (kcal/mole) given in ()] via zwitterions , leaving the concerted [$\pi^4_s + \pi^2_s$] ³ pathway ? As yet lack of stereospecificity is the only unequivocal proof of a two or more step mechanism . In a 1:1 molar ratio 2a , 2e or 2k , and pure E- resp. Z-olefins [dimethyl fumarate 1a or maleate 3a (CHCl₃ , 39.1) ⁴ , methyl cinnamate 1b (chlorobenzene , 37.5) ⁵ , 4-nitrophenyl cinnamate 1c (toluene , 33.9) ⁵ , β -nitrostyrene 1d (CH₂Cl₂ , 41.1)] only yielded the permitted stereoisomers of a concerted reaction . To elucidate the surprising lack of stereospecificity observed after the addition of 1d to 2a [we isolated formerly 6d and 7d ⁷] meant to solve three problems : 1) Stereochemistry of further 1,3-dipolar cycloadditions using highly polar reactants ? 2) New non stereospecific pathways ? 3) Catalysed epimerisation at C-2 during work-up ?

1) The pyrazolid-3-one azomethinimines 2f , 2g [in CH₂Cl₂] and 2h [in chlorobenzene] with E- β -nitrostyrene 1d [molar ratio 1:1] only yielded the "permitted" 6f , 6g and 6h ² . 6g easily epimerized to 7g [CH₂Cl₂ , basic alumina , 20°C ; 60% 6g + 40% 7g] ² . 1d added to HUISGEN's highly polar [$\mu(\underline{8h})=6.7$ D , dioxane ⁸] azomethinimines 8h and 9h [molar ratio 1:1 , 130h reflux , CH₂Cl₂] only to the "permitted" 2-cyano-3-phenyl-4ref,trans-nitro-pyrazolidin-5-spiro-9'-fluorenes 10h [m.p. 170-175°C , dec. , $\delta(H-3)=5.93$, $\delta(H-4)=5.95$, $J_{34}=8.8$, 200 MHz ; proof by X ray analysis] and 11h [m.p. 177-183°C , dec. , $\delta(H-3)=5.94$, $\delta(H-4)=5.95$, $J_{34}=9.0$, 200 MHz] ⁹ .



	R ¹	R ²	R ³	R ⁴	from <u>1</u>	μ [D] of <u>1</u>
<u>a</u>	H	Ph	CO ₂ Me	CO ₂ Me	<u>5</u> + <u>6</u> ^{a)}	2.25, CCl ₄
<u>b</u>	H	Ph	Ph	CO ₂ Me	<u>5</u> + <u>6</u>	1.93, diox.
<u>c</u>	H	Ph	Ph	CO ₂ -C ₆ H ₄ -NO ₂ -4	<u>4</u> ^{b)} + <u>5</u> + <u>6</u>	6.0, diox.
<u>d</u>	H	Ph	Ph	NO ₂	<u>5</u> ^{c)} + <u>6</u>	4.27, diox.
<u>e</u>	H	C ₆ H ₄ -Cl-4	C ₆ H ₄ -Cl-4	NO ₂	<u>6</u>	
<u>f</u>	Me	Me	Ph	NO ₂	<u>5</u> = <u>6</u>	
<u>g</u>	cyclohexylidene		Ph	NO ₂	<u>5</u> = <u>6</u>	
<u>h</u>	fluorenylidene		Ph	NO ₂	<u>5</u> = <u>6</u>	
<u>i</u>	H	Ph	C ₆ H ₄ -OMe-4	NO ₂	<u>6</u> ^{d)}	
<u>k</u>	H	C ₆ H ₄ -OMe-4	Ph	NO ₂	<u>6</u> ^{e)}	

All ¹H nmr data : if not given otherwise , 100 MHz , CDCl₃ , TMS int. , δ [ppm] , J [Hz] , calc. ABC systems . i.r. : [cm⁻¹] in KBr .

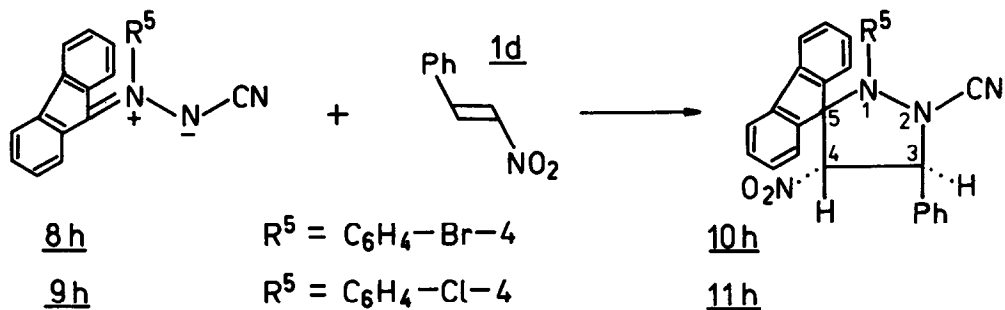
a) Numbering at poss. 1 and 3 of 5a , 6a and 7a must be exchanged ⁴ ; 2a and dimethyl maleate (3a , R³=R⁴=CO₂Me ; μ =2.48 D , CCl₄) only yield 7 (R¹=H , R²=Ph) and 7 (R¹=Ph , R²=H) ⁴ .

b) 4c : c. 5% of yield , m.p. 178-181°C ; δ (H-1)=3.65 , δ (H-2)=3.93 , δ (H-3)=5.10 , J₁₂=10.0 , J₁₃=0.0 , J₂₃=7.5 .

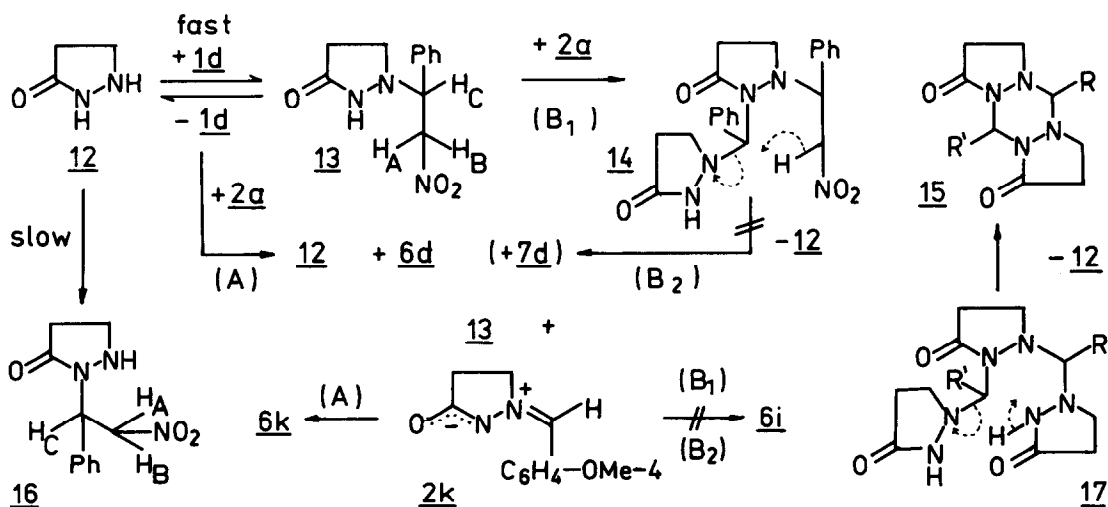
c) We detected 5d in a new type of reaction [2a + MeNO₂ ⁶] , then after addition of 1d to 2a [less than 1% 5d] : m.p. 105-108°C ; δ (H-1)=4.18 , δ (H-2)=5.01 , δ (H-3)=5.60 , J₁₂=8.0 , J₁₃=0.0 , J₂₃=4.5 ; ν (C=O) = 1707 .

d) 6i : m.p. 166-172°C (dec.=retro rc.) ; δ (H-1)=4.29 , δ (H-2)=5.22 , δ (H-3)=5.53 , J₁₂=6.8 , J₁₃=0.0 , J₂₃=2.5 .

e) 6k : m.p. 171-177°C (dec.=retro rc.) ; δ (H-1)=4.28 , δ (H-2)=5.22 , δ (H-3)=5.62 , J₁₂=6.5 , J₁₃=0.0 , J₂₃=2.9 ; for assignment use δ (H-3) !



2) We found 13 [m.p. 99-104°C (dec.=retro rc.), $\delta_A=4.63$, $\delta_B=4.92$, $\delta_C=4.52$, $J_{AB}=-13.0$, $J_{AC}=6.0$, $J_{BC}=8.5$] as kinetically controlled MICHAEL adduct of E- β -nitrostyrene 1d to pyrazolid-3-one 12 . 13 via 1d and 12 slowly rearranges to the thermodynamically controlled 16 [m.p. 87-88°C , $\delta_A=4.94$, $\delta_B=5.30$, $\delta_C=5.91$, $J_{AB}=-13.0$, $J_{AC}=4.5$, $J_{BC}=10.5$, CD_3OD] . Pyrazolid-3-one azomethinimines 2 mostly contain more or less traces of pyrazolid-3-one 12 ! We excluded the non stereospecific pathway B_1/B_2 via a MANNICH-like addition of 2 to 13 (B_1) and elimination of pyrazolid-3-one 12 from 14 (B_2) : 13 with 1-(4-methoxy-benzylidene)pyrazolid-3-one-azomethinimine 2k yielded 12 , and 6k (path A) , i.e. the 1,3-dipolar cyclo-adduct of E- β -nitrostyrene 1d to 2k ¹⁰ , whereas via B_1/B_2 the isomer 6i should have been formed . As by-products we isolated "thermal dimers" 15 of



pyrazolid-3-one azomethinimines 2, above all a mixed 15 with $R=C_6H_4-OMe-4$ and $R'=Ph$, formed via the MANNICH-like adduct 17 according to our mechanism of the thermal 1,3-dipole dimerisation 11.

Results 1) + 2), and later on valuable discussions with Prof. Rolf Huisgen forced us into further studies on epimerisation at C-2, especially of 6d, which formerly was isolated after elution from Kieselgel (KG) 40 MERCK (0.06-0.2 mm) with CH_2Cl_2 ⁷. After pouring 1.00 g of 6d in 10 ml CH_2Cl_2 on a column with 40 g KG 40, followed by elution with CH_2Cl_2 or CH_2Cl_2 and MeOH at 20°C no 7d was detected by ¹H nmr. If 7 mg of pyrazolid-3-one 12 [purified by Kugelrohr distillation at 175-200°C, 1 - 0.5 Pa] were added, similar elution gave 91% 6d and 9% 7d (by ¹H nmr). 1d and 2a, which can be purified by trituration with MeOH, according to ⁷ but without passing a KG 40 column only gave 6d (¹H nmr control). Thus, a combination of surprisingly small amounts of pyrazolid-3-one 12, sparingly soluble in CH_2Cl_2 , and KG 40 catalyses epimerisation at C-2 of 6d, generating 7d under very mild conditions as an artefact.

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